

EXAMEN OTU 06: Basic structural elucidation applied to natural and synthetic compounds

Session: 1st (2021-2022)

May 11th, 2022

1 hour

NMR

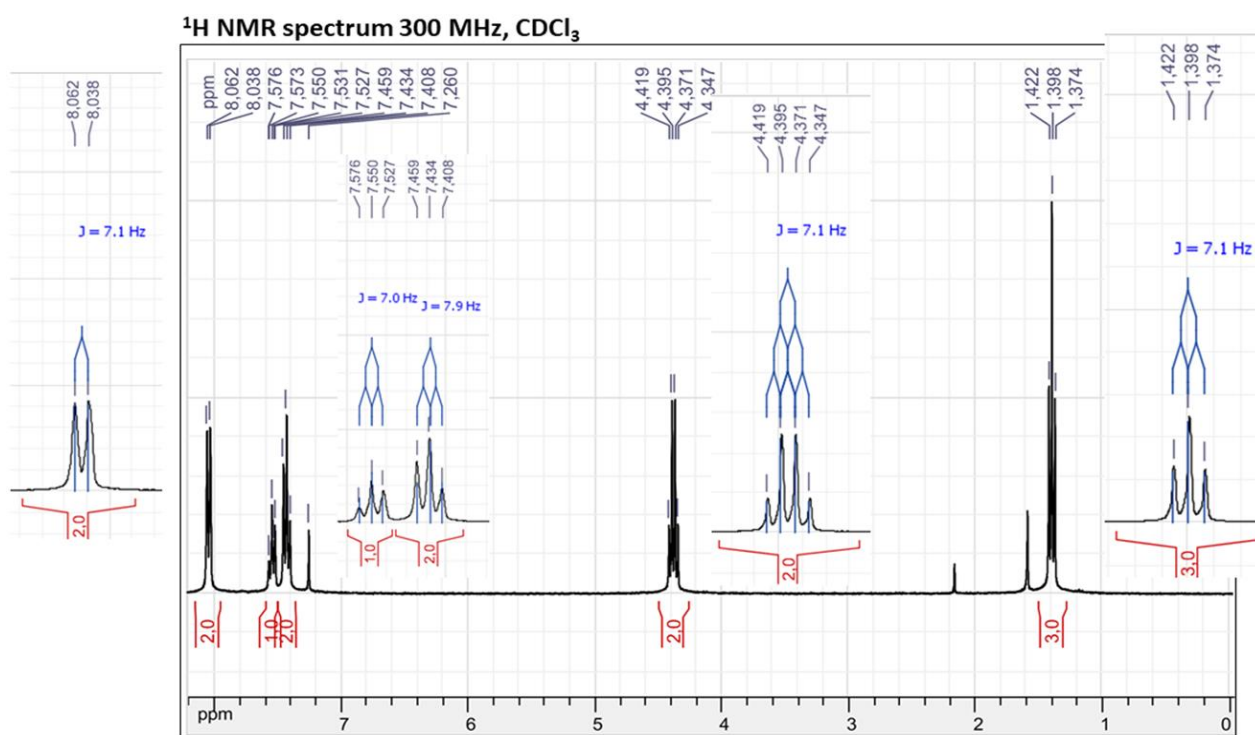
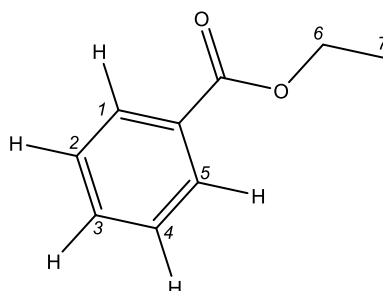


Figure 1: ¹H NMR spectrum of molecule A 300 MHz in CDCl₃ (peak-picking in ppm)

The provided ¹H NMR spectrum (Figure 1) is associated with the compound disclosed below.



Assign the ^1H NMR chemical shifts to the appropriate protons of the structure using the numbering system used above.

Summarize your data as a table compiling the integrations, multiplicities and different coupling constants.

MS

The electronic impact (EI) mass spectrum of molecule A is reported on the Figure 2. As this spectrum is constituted of many fragments, this exercise will focus on the most important fragments that are reported in the table.

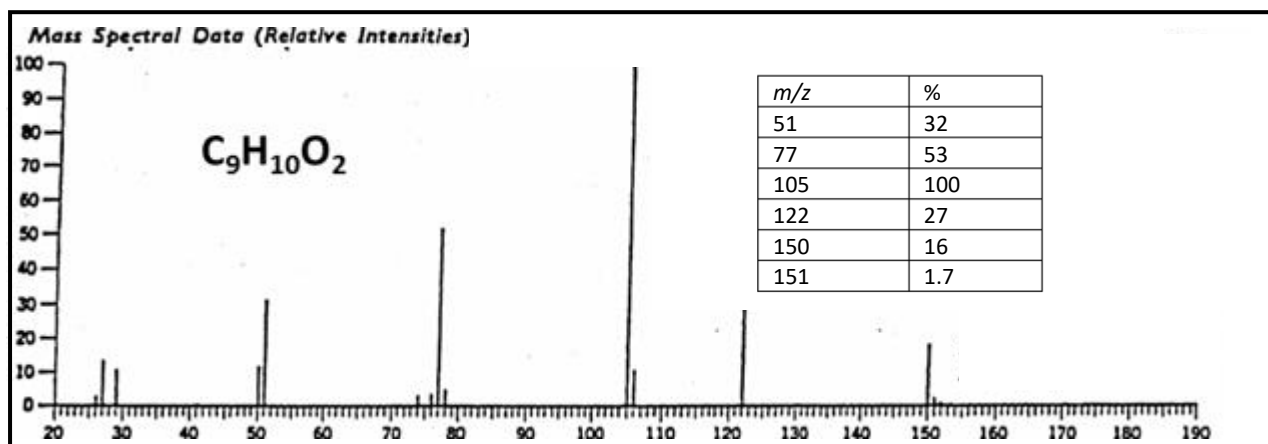


Figure 2: Mass spectrum of molecule A

1. Indicate which peak is the base peak and which is the molecular ion peak.
2. Indicate the nature of the ions at $m/z = 105$, 77 and 51 and detail the mechanism of their formation.
3. The formation of the ion at $m/z = 122$ is the result of a transposition mechanism, the neutral fragment having the chemical formula C_2H_4 . Explain the mechanism of its formation.