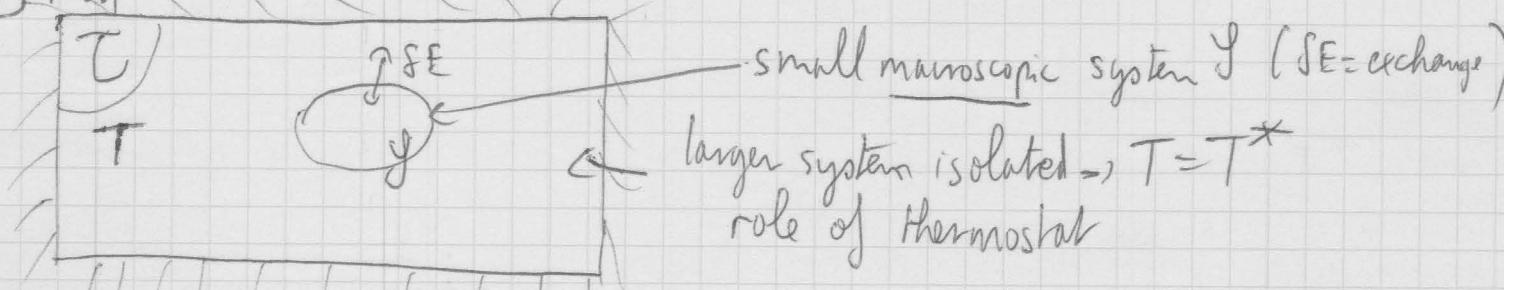


## chap 5 : Canonical ensemble

Intro : So far, we started with the isolated system  $\Rightarrow E$  fixed microcanonical  
 $\Rightarrow$  then we release some constraints between 2 isolated systems  
 The goal was to look at the final state reached when we release a constraint. However in realistic systems, the energy is not fixed  
 What is fixed then?

### I] The canonical ensemble

1st) Proof



$\Rightarrow$  Under these conditions, what is the distribut<sup>o</sup> of the microstates?

$$E_{\text{tot}} = E_T + E_Y = E_T + E \quad \text{fixed.}$$

Let us fix  $E$ . # microstates =  $\mathcal{N}_T(E) \cdot \mathcal{N}_Y(E_{\text{tot}} - E)$

$$\mathcal{N}_{T \otimes Y}(E_{\text{tot}}) = \sum_E \mathcal{N}_T(E) \mathcal{N}_Y(E_{\text{tot}} - E)$$

Here we neglect weak interactions at the border between  $Y$  and  $T$   
 (this is justified for macroscopic systems)

The proba distribut<sup>o</sup> of a state  $f \in Y$  is given by the total nb  
 of states in  $T \otimes Y$  when  $Y$  being in state  $f$

$$\text{For fixed } E_f \quad P_f = \frac{\mathcal{N}_T(E_{\text{tot}} - E_f)}{\sum_{E_f} \mathcal{N}_T(E_{\text{tot}} - E_f) \mathcal{N}_Y(E_f)} \propto e^{-\frac{1}{k_B} S_c^*(E_{\text{tot}} - E_f)}$$

$\uparrow$  Entropy

$$\text{However } S_c^*(E_{\text{tot}} - E_f) \approx S_c^*(E_{\text{tot}}) - E_f \frac{\partial S_c^*}{\partial E_f} \Big|_{E_{\text{tot}}} + O(E_f^2)$$

$$S_{\bar{T}}^*(E_{\text{tot}} - E_{\text{f}}) = S_{\bar{T}}^*(E_{\text{tot}}) - \frac{E_{\text{f}}}{\bar{T}} + O(E_{\text{f}}^2)$$

Here  $\boxed{T = T_{\bar{T}}^*}$  microcanonical temp of larger system.

$$p_l^c = \frac{1}{Z} e^{-\beta E_l}$$

with

$$\beta = \frac{1}{k_B T}$$

normalization:

$$Z = \sum_l e^{-\beta E_l}$$

canonical partition function

The thermostat is only encoded through  $\beta$

## 2) Another proof

let us use the maximized entropy principle

$$S = -k_B \sum_l p_l \ln p_l \quad \text{with} \quad \sum_l p_l = 1$$

$\{l\}$  labels the ministates of  $\mathcal{S}$  in contact with a thermostat  $\bar{T}$

We allow exchange of energy of  $\mathcal{S}$  with  $\bar{T}$  such that  $\langle E \rangle$  is fixed.

$$\sum_l E_l p_l = \langle E \rangle = E$$

$$\text{Consider } f(\{p_l\}) = -k_B \sum_l p_l \ln p_l \Leftrightarrow \alpha \sum_l (p_l - 1) + \beta k_B (\sum_l p_l E_l - E)$$

$$\frac{\partial f}{\partial p_l} = 0 \Rightarrow -k_B \ln p_l - k_B - \alpha - \beta k_B E_l = 0$$

$$\Rightarrow p_l = \frac{e^{-\beta E_l}}{1 + \alpha} \quad \alpha, \beta \in \mathbb{R}$$

$$\sum_l p_l = 1 \Rightarrow e^{1+\alpha} = \sum_l e^{-\beta E_l} \Rightarrow p_l = \frac{e^{-\beta E_l}}{\sum_l e^{-\beta E_l}} \leq Z$$

$$\text{However } p_l? \quad S = -k_B \sum_l p_l \ln p_l$$

$$= -k_B \sum_l p_l (-\beta E_l - \ln Z) = \beta k_B \langle E \rangle + k_B \ln Z$$

$$\frac{\partial S}{\partial \langle E \rangle} = \frac{1}{T} = \beta k_B \Rightarrow \beta = \frac{1}{k_B T} \quad ||$$

Here  $T$  is not really the microcanonical temperature but more some average  $T$  which is the thermostat temp. if in contact with  $\bar{T}$

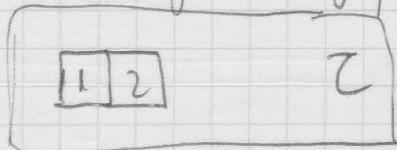
(3)

### 3) Helmholtz free energy and extensivity

$$Z \equiv e^{-\beta F_c} \Leftrightarrow F_c \equiv -k_B T \ln Z(T, V, N) //$$

F will be interpreted as the generating function of thermodynamic properties

F is extensive



If we neglect interactions between 1 and 2  $\{f_{l_1, l_2}\} \cong \{f_{l_1}\} \otimes \{f_{l_2}\}$

Then  ~~$E$~~   $E_{l_1, l_2} = E_1 + E_2$

$$Z = \sum_{l_1, l_2} e^{-\beta(E_{l_1} + E_{l_2})} = \left( \sum_{l_1} e^{-\beta E_{l_1}} \right) \left( \sum_{l_2} e^{-\beta E_{l_2}} \right) = Z_1 Z_2$$

$$\Rightarrow F = -k_B T \ln Z = F_1 + F_2$$

We can generalize it if  $\mathcal{Y} = \mathcal{Y}_1 \otimes \mathcal{Y}_2 \otimes \dots \otimes \mathcal{Y}_N$

then  $E_{l_1, l_2, \dots, l_N} = \sum_{i=1}^N \epsilon_{l_i}$

same reasoning  $Z = \prod_{i=1}^N z_i \Rightarrow F = -k_B T \sum_{i=1}^N \ln z_i$

$$\Rightarrow F = \sum_{i=1}^N F_i //$$

If the subsystems are distinguishable and identical  $Z = z^N \Rightarrow F = -N k_B T \ln z = N f$

extensivity:  $F(T, dV, dN) = dF(T, V, N)$

which implies  $F(T, V, N) = N f\left(T, \frac{V}{N}\right)$

F depends on only 2 variables!

↑ intensive      ↓ intensive

### 4) Average energy and higher cumulants

$$\bar{E}^c = \sum_{\ell} p_{\ell}^c E_{\ell} = \frac{1}{Z} \sum_{\ell} E_{\ell} e^{-\beta E_{\ell}} = \frac{1}{Z} \frac{d}{d\beta} \sum_{\ell} e^{-\beta E_{\ell}}$$

$$\Rightarrow \bar{E}^c = -\frac{1}{Z} \frac{dZ}{d\beta} = -\frac{d \ln Z}{d\beta}$$

$$\langle E^2 \rangle_c^* = \bar{E}^{2c} = \frac{1}{Z} \sum_{\ell} E_{\ell}^2 e^{-\beta E_{\ell}} = \frac{1}{Z} \frac{d^2}{d\beta^2} \left( \sum_{\ell} e^{-\beta E_{\ell}} \right)$$

$$= \frac{1}{Z} \frac{d^2 Z}{d\beta^2}$$

$$\begin{aligned} \langle \sigma_E \rangle^2 &= \langle E^2 \rangle_c^* - \langle E \rangle_c^2 = \frac{1}{Z} \frac{d^2 Z}{d\beta^2} - \frac{1}{Z^2} \left( \frac{dZ}{d\beta} \right)^2 \\ \text{var}(E) &= \left( -\frac{d}{d\beta} \right)^2 \ln Z = -\frac{d}{d\beta} \bar{E}^c \end{aligned}$$

$$\text{var}(E) = \sigma_E^2 = -\frac{d\bar{E}^c}{d\beta} = \left( -\frac{d}{d\beta} \right) \ln Z$$

In usual thermodynamics,  $C_V^h = \left( \frac{\partial E}{\partial T} \right)_{N,V}$

generating function  
of the cumulants

We have shown that the thermo def and  $\mu$ -canonical def make sense.

$$C_V^c = \frac{\partial \bar{E}^c}{\partial T} = -k_B \beta^2 \frac{\partial \bar{E}^c}{\partial \beta} = \frac{1}{k_B T^2} \sigma_E^2$$

$$C_V = \frac{1}{k_B T^2} \sigma_E^2$$

↑                      ↓  
thermo                fluctuations

We have established a relation between the specific heat and the fluctuations of energy of the system. (particular type of a more general relation which relates the fluctuations of the system to the way it dissipates (exchange energy with the outside))

(5)

## 5) Application to the paramagnetic crystal

$\mu$ -canonical

$$N = N_+ + N_-$$

$$E = (-N_+ + N_-)\varepsilon_B$$

$$\Rightarrow N^\pm = \frac{1}{2} \left( N \mp \frac{\varepsilon_B}{k_B T} \right)$$

$$\Omega = \frac{N!}{N_+! N_-!} \Rightarrow S^* = k_B \ln \Omega$$

$$\frac{1}{T^*} = \frac{\partial S^*(E)}{\partial E} = \frac{k_B}{2\varepsilon_B} \ln \frac{N\varepsilon_B - E}{N\varepsilon_B + E}$$

Invert this relation

$$2\beta^* \varepsilon_B = \ln \frac{N\varepsilon_B - E}{N\varepsilon_B + E}$$

$$\hookrightarrow E = -N\varepsilon_B \tanh \beta^* \varepsilon_B$$

Let us compare the two results: They agree if:

$E$	$=$	$\bar{E}^c$
$T^*$	$=$	$T$

Basically we go from  $\mu$ -canonical to canonical by replacing  $E$  by  $\bar{E}^c$  and  $T^*$  by  $T$ .

## 6) Entropy of a canonical ensemble

$$S = -k_B \sum_l p_l^c \ln p_l^c = -k_B \sum_l p_l^c (-\ln Z - \mu_E l)$$

$$= k_B \ln Z + \frac{1}{T} \bar{E}^c = \frac{\bar{E}^c - F}{T}$$

$S = \frac{\bar{E}^c - F}{T}$
-------------------------------

$$F = -k_B T \ln Z \Rightarrow \frac{dF}{dT} = -k_B \ln Z + \frac{1}{Z} \frac{dZ}{dT} (-k_B T)$$

$$\text{and } \frac{d \ln Z}{dT} = \frac{d \ln Z}{d\beta} \frac{d\beta}{dT} = -\frac{1}{k_B T^2} \frac{d \ln Z}{d\beta} = \frac{\bar{E}^c}{k_B T^2}$$

$$\text{Therefore } \frac{dF}{dT} = -k_B \ln Z - k_B T \frac{\bar{E}^c}{k_B T^2} = -k_B \ln Z - \frac{\bar{E}^c}{T} = -S$$

Thus  $\boxed{S = -\frac{dF}{dT}}$        $\text{Also } \text{Ex } C_V = T \frac{dS}{dT} = -T \frac{\partial F}{\partial T}$

We have found  $S = \frac{\bar{E}^c - F}{T} \Rightarrow \boxed{F = \bar{E}^c - TS}$

This is fully analogous to thermo where  $F = U - TS$

RQ: In thermodynamics  $\Delta S \geq \int \frac{dQ}{T}$

if T fixed  $\Delta S \geq \frac{Q}{T} \Leftrightarrow Q \leq T \Delta S$

$T \Delta S$  defines some upper limit to the heat received.

$\Delta F = \Delta E - T \Delta S = \Delta U - T \Delta S = W + Q - T \Delta S \leq W$

Availability  $= -W \leq -\Delta F$  upper limit of the work which is available

The relations between  $F^c$  and  $S^c$  is called a Legendre transform

$S^c$  is the derivative of  $F$  with respect to  $T$

$(T, S)$  are conjugate variable

$\Rightarrow$  Mathematical  
inner SVP

In thermodynamics, we go from thermodynamical potential to another one by a Legendre transform.

$$dE = T dS - pdV + \mu dN$$

$$F = E - TS \quad dF = dE - T dS - SdT$$

$$\Leftrightarrow dF = -SdT - pdV + \mu dN$$

$$\Leftrightarrow S = -\left(\frac{\partial F}{\partial T}\right)_{V, N}$$

(7)

### 7) Relations with Thermodynamics (1<sup>st</sup> and 2<sup>nd</sup> principle)

$$d\bar{E}^c = d\left(\sum_l p_l E_l\right) = \sum_l dP_l E_l + \sum_l dE_l P_l$$

Let us consider a reversible transformation

$$\text{Since } U^h \xrightarrow{\text{rev}} \bar{E}^c \Rightarrow dU = \delta Q + \delta W$$

could we identify  $d\bar{E}^c = \underbrace{\sum_l E_l dP_l}_{\delta Q} + \underbrace{\sum_l p_l dE_l}_{\delta W}$  ?

Not obvious a priori

$$\begin{aligned} S &= -k_B \sum_l p_l \ln p_l = dS = -k_B \sum_l (\ln p_l + 1) dP_l \\ &\quad = k_B \sum_l (-1 + \ln 2\pi + \beta E_l) dP_l \\ \Downarrow \quad \sum_{\substack{dP_l=0 \\ ||}} \quad dS &= k_B \beta \sum_l E_l dP_l = \frac{1}{T} \sum_l E_l dP_l \\ \Rightarrow \quad \delta Q &= \underbrace{\sum_l E_l dP_l}_{\text{rev}} \quad || \end{aligned}$$

Therefore  $\delta W^m = \sum_l p_l dE_l \quad ||$

This provides a microscopic interpretation of the 1<sup>st</sup> principle  
A We use a transformation which is reversible

Ex: Let us come back to our paramagnetic crystal.

$$F = -k_B T \ln(2 \cosh(\beta E_B)) \Rightarrow S = -\frac{\partial F}{\partial T} = N k_B \left[ \ln(2 \cosh(\beta E_B)) - \beta E_B \tanh(\beta E_B) \right]$$

Ex: Compare it to  $\mu$ -canonical

$$H = -\mu_B \sum_{i=1}^N s_i \text{ with } s_i = \pm 1 \Rightarrow M = \mu_B \sum_i s_i = \overline{M}^C = \mu_B N \tanh(\beta E_B)$$

$$\overline{E}_B = \frac{\mu_B \sqrt{B}}{2} \stackrel{!}{=} \mu_B B \quad \overline{m}^C = \mu_B \tanh(\beta \overline{E}_B); \quad \chi'(B) = \frac{\partial \overline{m}^C}{\partial B} = \beta \mu_B^2 (1 - \tanh^2(\beta \overline{E}_B))$$

$$\chi(0) = \beta \mu_B^2 = \frac{\mu_B^2}{k_B T} \xrightarrow{T \rightarrow 0} \infty \neq \lim_{T \rightarrow 0} \chi(B) \sim \frac{\mu_B^2}{k_B T} e^{-\frac{2\mu_B B}{k_B T}} \xrightarrow{B \rightarrow 0} 0$$

## 8) About the third principle.

7'

The 3<sup>rd</sup> principle is actually an abuse of language for He Nerst theorem

$$S(T) \xrightarrow{T \rightarrow 0} 0$$

However by def  $\begin{cases} S^*(T) \rightarrow k_B \ln D_0 \\ S^c(T) \xrightarrow{T \rightarrow 0} \end{cases}$  where  $D_0$  is the ground state degeneracy

However one may reasonably expect that the degeneracy of the ground state to be non extensive. such that

$$\lim_{T \rightarrow 0} \frac{1}{N} S(T) \rightarrow 0 \quad \text{Nerst theorem}$$

However, we know systems where the degeneracy of the ground state can be extremely large such that  $\lim_{T \rightarrow 0} \frac{S(T)}{N} = \text{finite}$

Ex: Glasses, spin liquids, ...