

Appendix B Common Fragment Ions

All fragments listed bear +1 charges. To be used in conjunction with Appendix C. Not all members of homologous and isomeric series are given. The list is meant to be suggestive rather than exhaustive. Appendix II of Hamming and Foster

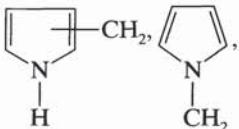


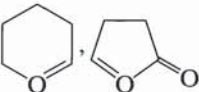


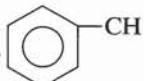
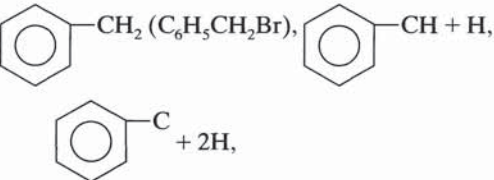
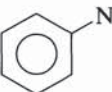
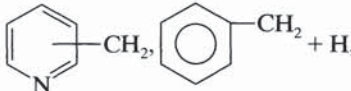
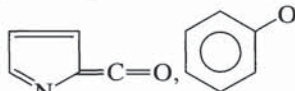
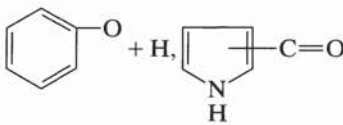
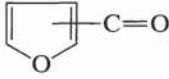
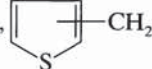
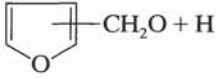
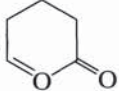
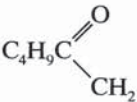
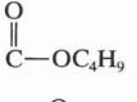
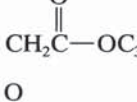
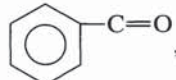
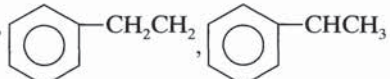
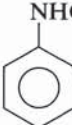
(1972), Table A-7 of McLafferty's (1993) interpretative book, and the high-resolution ion data of McLafferty (1982) are recommended as supplements. Structural inferences are listed in parentheses.

<i>m/z</i> Ions ^a	(Structural Inference)
14	CH ₂
15	CH ₃
16	O
17	OH
18	H ₂ O, NH ₄
19	F, H ₃ O
26	C≡N, C ₂ H ₂
27	C ₂ H ₃
28	C ₂ H ₄ , CO, N ₂ (air), CH=NH
29	C ₂ H ₅ , CHO
30	CH ₂ NH ₂ (RCH ₂ NH ₂), NO
31	CH ₂ OH (RCH ₂ OH), OCH ₃
32	O ₂ (air)
33	SH, CH ₂ F
34	H ₂ S
35	Cl (³⁷ Cl at 37)
36	HCl (H ³⁷ Cl at 38)
39	C ₃ H ₃
40	CH ₂ C=N, Ar(air)
41	C ₃ H ₅ , CH ₂ C=N + H, ^a C ₂ H ₂ NH
42	C ₃ H ₆ , C ₂ H ₂ O
43	C ₃ H ₇ , CH ₃ C=O, CH ₃ C=OG, (G = R, Ar, NH ₂ , OR, OH), C ₂ H ₅ N
44	$\begin{array}{c} \text{H} \\ \\ \text{CH}_2\text{C}=\text{O} \end{array} + \text{H}$ (Aldehydes, McLafferty rearrangement), CH ₃ CHNH ₂ , CO ₂ , NH ₂ C=O (RC=ONH ₂), (CH ₃) ₂ N
45	$\begin{array}{c} \text{CH}_3 \\ \\ \text{CHOH} \end{array}$, CH ₂ CH ₂ OH, CH ₂ OCH ₃ (RCH ₂ OCH ₃), $\begin{array}{c} \text{O} \\ \\ \text{C}-\text{OH} \end{array}$, CH ₃ CH-O + H (CH ₃ CHOHR)
46	NO ₂
47	CH ₂ SH (RCH ₂ SH), CH ₃ S
48	CH ₃ S + H
49	CH ₂ Cl (CH ₂ ³⁷ Cl at 51)
51	CHF ₂ , C ₄ H ₃
53	C ₄ H ₅
54	CH ₂ CH ₂ C≡N
55	C ₄ H ₇ , CH ₂ =CHC=O
56	C ₄ H ₈
57	C ₄ H ₉ , C ₂ H ₅ C=O
58	$\begin{array}{c} \text{O} \\ \\ \text{CH}_3-\text{C} \\ \\ \text{CH}_2 \end{array} + \text{H}$, C ₂ H ₅ CHNH ₂ , (CH ₃) ₂ NCH ₂ , C ₂ H ₅ NHCH ₂ , C ₂ H ₂ S
59	(CH ₃) ₂ COH, CH ₂ OC ₂ H ₅ , $\begin{array}{c} \text{O} \\ \\ \text{C}-\text{OCH}_3 \end{array}$ (RCO ₂ CH ₃), NH ₂ C=O + H, CH ₃ OCHCH ₃ , CH ₃ CHCH ₂ OH, $\begin{array}{c} \text{CH}_2 \\ \\ \text{C}_2\text{H}_5\text{CHOH} \end{array}$
60	$\begin{array}{c} \text{O} \\ \\ \text{CH}_2\text{C} \\ \\ \text{OH} \end{array} + \text{H}$, CH ₂ ONO
61	$\begin{array}{c} \text{O} \\ \\ \text{CH}_3\text{C}-\text{O} \end{array} + 2\text{H}$, CH ₂ CH ₂ SH, CH ₂ SCH ₃
65	C ₅ H ₅
66	 ≡ C ₅ H ₆ , H ₂ S ₂ (RSSR)
67	C ₅ H ₇
68	CH ₂ CH ₂ CH ₂ C≡N
69	C ₅ H ₉ , CF ₃ , CH ₃ CH=CHC=O, CH ₂ =C(CH ₃)C=O
70	C ₅ H ₁₀
71	C ₅ H ₁₁ , C ₃ H ₇ C=O
72	$\begin{array}{c} \text{O} \\ \\ \text{C}_2\text{H}_5\text{C} \\ \\ \text{CH}_2 \end{array} + \text{H}$, C ₃ H ₇ CHNH ₂ , (CH ₃) ₂ N=C=O, C ₂ H ₅ NHCHCH ₃ and isomers
73	Homologs of 59, (CH ₃) ₃ Si
74	$\begin{array}{c} \text{O} \\ \\ \text{CH}_2-\text{C}-\text{OCH}_3 \end{array} + \text{H}$
75	$\begin{array}{c} \text{O} \\ \\ \text{C}-\text{OC}_2\text{H}_5 \end{array} + 2\text{H}$, $\begin{array}{c} \text{O} \\ \\ \text{C}_2\text{H}_5\text{CO} \end{array} + 2\text{H}$, CH ₂ SC ₂ H ₅ , (CH ₃) ₂ CSH, (CH ₃ O) ₂ CH, (CH ₃) ₂ SiOH
76	C ₆ H ₄ (C ₆ H ₅ X, C ₆ H ₄ XY)
77	C ₆ H ₅ (C ₆ H ₅ X)
78	C ₆ H ₅ + H
79	C ₆ H ₅ + 2H, Br (⁸¹ Br at 81)

Appendix B Continued

 m/z Ions^a

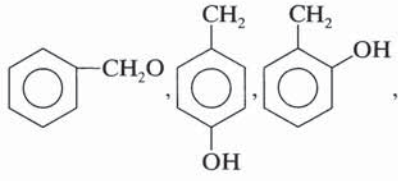
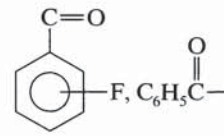
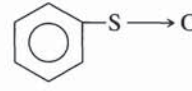
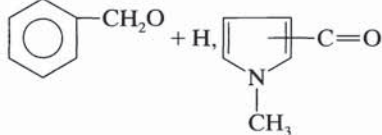
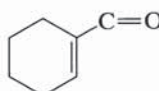
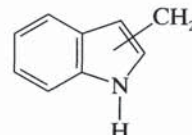
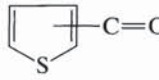
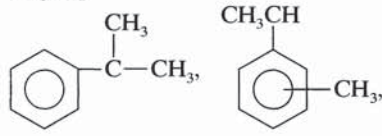
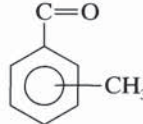
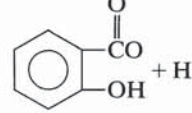
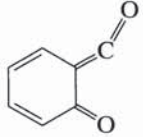
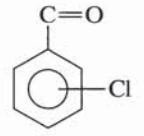
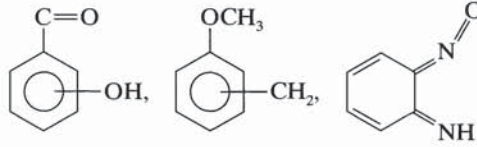
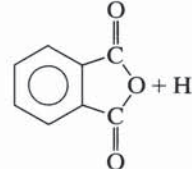
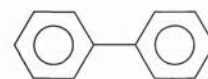
(Structural Inference)

- 80 
 $\text{CH}_3\text{SS} + \text{H}$, HBr (H^{81}Br at 82)
- 81 
 C_6H_9
- 82 $\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{C}\equiv\text{N}$, CCl_2 ($\text{C}^{35}\text{Cl}^{37}\text{Cl}$ at 84, C^{37}Cl_2 at 86), C_6H_{10}
- 83 C_6H_{11} , CHCl_2 ($\text{CH}^{35}\text{Cl}^{37}\text{Cl}$ at 85, $\text{CH}^{37}\text{Cl}_2$ at 87),

- 85 C_6H_{13} , $\text{C}_4\text{H}_9\text{C}=\text{O}$, CClF_2 ($\text{C}^{37}\text{ClF}_2$ at 87),

- 86 $\text{C}_3\text{H}_7\text{C}(=\text{O})\text{CH}_2 + \text{H}$, $\text{C}_4\text{H}_9\text{CHNH}_2$ and isomers
- 87 $\text{C}_3\text{H}_7\text{CO}$, homologs of 73, $\text{CH}_2\text{CH}_2\text{COCH}_3$

- 88 $\text{CH}_2-\text{C}(=\text{O})-\text{OC}_2\text{H}_5 + \text{H}$
- 89 $\text{C}-\text{OC}_3\text{H}_7 + 2\text{H}$, 
- 90 $\text{CH}_3\text{CHONO}_2$, 
- 91 
 $(\text{CH}_2)_4\text{Cl}$ [$(\text{CH}_2)_4^{37}\text{Cl}$ at 93], 
- 92 
 $\text{CH}_2 + \text{H}$,
- 93 CH_2Br ($\text{CH}_2^{81}\text{Br}$ at 95, RCH_2Br), C_7H_9 ,

 C_7H_9 (terpenes)
- 94 
 H ,
- 95 
- 96 $\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{C}\equiv\text{N}$
- 97 C_7H_{13} , 
- 98 
 $\text{CH}_2\text{O} + \text{H}$
- 99 C_7H_{15} , $\text{C}_6\text{H}_{11}\text{O}$, 
- 100 $\text{C}_4\text{H}_9\text{C}(=\text{O})\text{CH}_2 + \text{H}$, $\text{C}_5\text{H}_{11}\text{CHNH}_2$

- 101 $\text{C}-\text{OC}_4\text{H}_9$

- 102 $\text{CH}_2\text{C}(=\text{O})-\text{OC}_3\text{H}_7 + \text{H}$

- 103 $\text{C}-\text{OC}_4\text{H}_9 + 2\text{H}$, $\text{C}_5\text{H}_{11}\text{S}$, $\text{CH}(\text{OCH}_2\text{CH}_3)_2$
- 104 $\text{C}_2\text{H}_5\text{CHONO}_2$
- 105 
 $\text{C}_6\text{H}_5(\text{C}=\text{O})\text{G}$ [$\text{G} = \text{OH}$, OR , OAr , halogen, NH_2], 
- 106 NHCH_2


Appendix B Continued

m/z Ions^a

(Structural Inference)

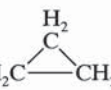
107		123	 $\text{F}, \text{C}_6\text{H}_5\text{C}(=\text{O})-\text{O} + 2\text{H}$
	$\text{C}_2\text{H}_4\text{Br}$ ($\text{C}_2\text{H}_4^{81}\text{Br}$ at 109)	125	
108		127	I
		128	HI
109		130	
111		131	$\text{C}_3\text{F}_5, \text{C}_6\text{H}_5-\text{CH}=\text{CH}-\text{C}(=\text{O})$
119	$\text{CF}_3\text{CF}_2,$ 	135	$(\text{CH}_2)_4\text{Br}$ [$(\text{CH}_2)_4^{81}\text{Br}$ at 137]
		138	 + H
120		139	
121	C_9H_{13} (terpenes) 	141	CH_2I (RCH_2I)
122	$\text{C}_6\text{H}_5\text{C}(=\text{O})-\text{O} + \text{H}, (\text{C}_6\text{H}_5\text{CO}_2\text{R})$	147	$(\text{CH}_3)_2\text{Si}=\text{O}-\text{Si}(\text{CH}_3)_3$
		149	 + H
		154	

^a Ions indicated as a fragment + *n*H (*n* = 1, 2, 3, . . .) are ions that arise via rearrangement involving hydrogen transfer.

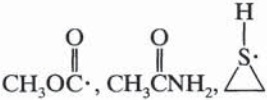
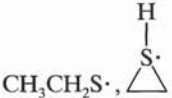
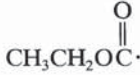
Appendix C Common Fragments Lost

This list is suggestive rather than comprehensive. It should be used in conjunction with Appendix B. Table 5-19 of Hamming and Foster (1972) and Table A-5 of McLafferty (1993) are

recommended as supplements. All of these fragments are lost as neutral species.

Molecular Ion Minus	Fragment Lost (Inference Structure)
1	H·
2	2H·
15	CH ₃ ·
16	O (ArNO ₂ , amine oxides, sulfoxides); ·NH ₂ (carboxamides, sulfonamides)
17	HO·
18	H ₂ O (alcohols, aldehydes, ketones)
19	F·
20	HF
26	CH≡CH, ·CH≡N
27	CH ₂ =CH·, HC≡N (aromatic nitrites, nitrogen heterocycles)
28	CH ₂ =CH ₂ , CO, (quinones) (HCN + H)
29	CH ₃ CH ₂ ·, (ethyl ketones, ArCH ₂ CH ₂ CH ₃), ·CHO
30	NH ₂ CH ₂ ·, CH ₂ O (ArOCH ₃), NO (ArNO ₂), C ₂ H ₆
31	·OCH ₃ (methyl esters), ·CH ₂ OH, CH ₃ NH ₂
32	CH ₃ OH, S
33	HS· (thiols), (·CH ₃ and H ₂ O)
34	H ₂ S (thiols)
35	Cl·
36	HCl, 2H ₂ O
37	H ₂ Cl (or HCl + H)
38	C ₃ H ₂ , C ₂ N, F ₂
39	C ₃ H ₃ , HC ₂ N
40	CH ₃ C≡CH
41	CH ₂ =CHCH ₂ ·
42	CH ₂ =CHCH ₃ , CH ₂ =C=O,  , NCO, NCNH ₂
43	C ₃ H ₇ · (propyl ketones, ArCH ₂ -C ₃ H ₇), CH ₃ C· (methyl ketones, CH ₃ C=O, where G = various functional groups), CH ₂ =CH-O·, (CH ₃ · and CH ₂ =CH ₂), HCNO
44	CH ₂ =CHOH, CO ₂ (esters, anhydrides), N ₂ O, CONH ₂ , NHCH ₂ CH ₃
45	CH ₃ CHOH, CH ₃ CH ₂ O· (ethyl esters), CO ₂ H, CH ₃ CH ₂ NH ₂
46	(H ₂ O and CH ₂ =CH ₂), CH ₃ CH ₂ OH, ·NO ₂ (ArNO ₂)
47	CH ₃ S·
48	CH ₃ SH, SO (sulfoxides), O ₃
49	·CH ₂ Cl
51	·CHF ₂
52	C ₄ H ₄ , C ₂ N ₂
53	C ₄ H ₅
54	CH ₂ =CH-CH=CH ₂
55	CH ₂ =CHCHCH ₃

Appendix C Continued

Molecular Ion Minus	Fragment Lost (Inference Structure)
56	$\text{CH}_2=\text{CHCH}_2\text{CH}_3$, $\text{CH}_3\text{CH}=\text{CHCH}_3$, 2CO
57	$\text{C}_4\text{H}_9\cdot$ (butyl ketones), $\text{C}_2\text{H}_5\text{CO}$ (ethyl ketones, $\text{EtC}=\text{OG}$, G = various structural units)
58	$\cdot\text{NCS}$, ($\text{NO} + \text{CO}$), CH_3COCH_3 , C_4H_{10}
59	$\text{CH}_3\text{OC}\cdot$, CH_3CNH_2 , \triangle 
60	$\text{C}_3\text{H}_7\text{OH}$, $\text{CH}_2=\text{C}(\text{OH})_2$ (acetate esters) ^a
61	$\text{CH}_3\text{CH}_2\text{S}\cdot$, \triangle 
62	(H_2S and $\text{CH}_2=\text{CH}_2$)
63	$\cdot\text{CH}_2\text{CH}_2\text{Cl}$
64	C_5H_4 , S_2 , SO_2
68	$\text{CH}_2=\overset{\text{CH}_3}{\text{C}}-\text{CH}=\text{CH}_2$
69	$\text{CF}_3\cdot$, $\text{C}_3\text{H}_9\cdot$
71	$\text{C}_5\text{H}_{11}\cdot$
73	$\text{CH}_3\text{CH}_2\text{OC}\cdot$ 
74	$\text{C}_4\text{H}_9\text{OH}$
75	C_6H_3
76	C_6H_4 , CS_2
77	C_6H_5 , CS_2H
78	C_6H_6 , CS_2H_2 , $\text{C}_3\text{H}_4\text{N}$
79	$\text{Br}\cdot$, $\text{C}_5\text{H}_5\text{N}$
80	HBr
85	$\cdot\text{CClF}_2$
100	$\text{CF}_2=\text{CF}_2$
119	$\text{CF}_3-\text{CF}_2\cdot$
122	$\text{C}_6\text{H}_5\text{COOH}$
127	$\text{I}\cdot$
128	HI

^a McLafferty rearrangement.